

Name _____

Chemistry

___/___/___

Chapter 12 Vocabulary - Define each of the following on a separate sheet of paper. All definitions must be hand written.

1. chemical equilibrium
2. common ion
3. common ion effect
4. complex ions
5. equilibrium constant
6. equilibrium position
7. Le Chatelier's principle
8. ligand
9. reversible reaction
10. solubility-product constant

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Chapter 13 Vocabulary - Define each of the following on a separate sheet of paper. All definitions must be hand written.

1. acid
2. acid anhydride
3. acid dissociation constant
4. alkaline solution
5. amphoteric
6. amphoteric
7. base
8. base anhydride
9. base dissociation constant
10. basic solution
11. Bronsted-Lowry acid
12. Bronsted-Lowry base
13. buffer
14. conjugate acid
15. conjugate base
16. diprotic acid
17. dissociation constant for water
18. end point
19. equivalence point
20. fatty acid
21. hydrogen-ion acceptor
22. hydrogen-ion donor
23. hydronium ion
24. hydroxide ion
25. Lewis acid
26. Lewis base
27. monoprotic acid
28. indicator
29. neutralization reaction
30. pH
31. pOH
32. salt
33. self-ionization
34. strong acid
35. strong base
36. titration
37. triprotic acid
38. weak acid
39. weak base