

	Solute	+	Solvent	=	Solution
<b>volume</b>	% volume		% volume		% volume Molarity
<b>convert between volume and mass using density</b>					
<b>mass</b>	% mass		% mass molality (in kg)		% mass
<b>convert between mass and moles using the gram formula mass</b>					
<b>moles</b>	mole fraction Molarity molality		mole fraction		mole fraction

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